



TABLE 1. SUMMARY OF THE DATED SAMPLES.

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### Results of Radiocarbon Dating

Dear sir, please find enclosed the results of the radiocarbon dating of the samples you submitted to CEDAD (AMS and radiocarbon dating facility, University of Lecce, Italy) and listed in Table 1.

<b>Sample ID</b>	<b>CEDAD Code</b>	<b>Provenience</b>
20_0007_0001	LTL20328A	
20_0007_0002	LTL20329	
20_0007_0004	LTL20330A	
20_0007_0006	LTL20331A	
20_0007_0007	LTL20332A	
20_0007_0008	LTL20333A	

Macro contaminants were removed from the samples by mechanical handpicking under optical microscope. The selected portion of the samples was treated in order to chemically remove any possible source of contamination.

The purified sample material was then converted to carbon dioxide by combustion in sealed quartz tubes. The obtained carbon dioxide was converted at 550°C into graphite by using ultrahigh purity Hydrogen as reducing medium and 2 mg iron powder as catalyst.



TABLE 2. MEASURED VALUES.

The sample yielded enough graphite to allow an accurate determination of the radiocarbon age by the accelerator mass spectrometer.

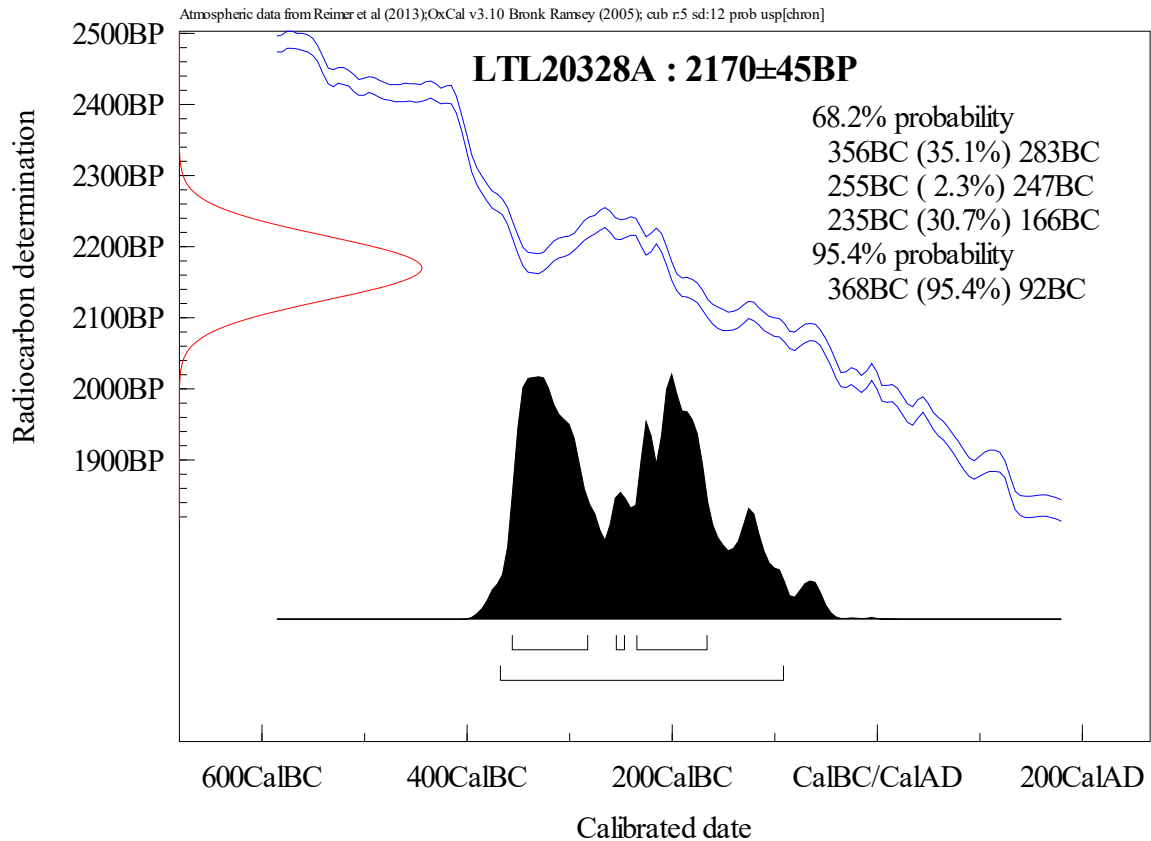
The radiocarbon concentrations have been determined in the accelerator mass spectrometer by comparing the  $^{12}\text{C}$ ,  $^{13}\text{C}$  currents and the  $^{14}\text{C}$  counts obtained from the samples with those obtained from standard materials supplied by IAEA (International Atomic Energy Agency) and NIST (National Institute of Standard and Technology).

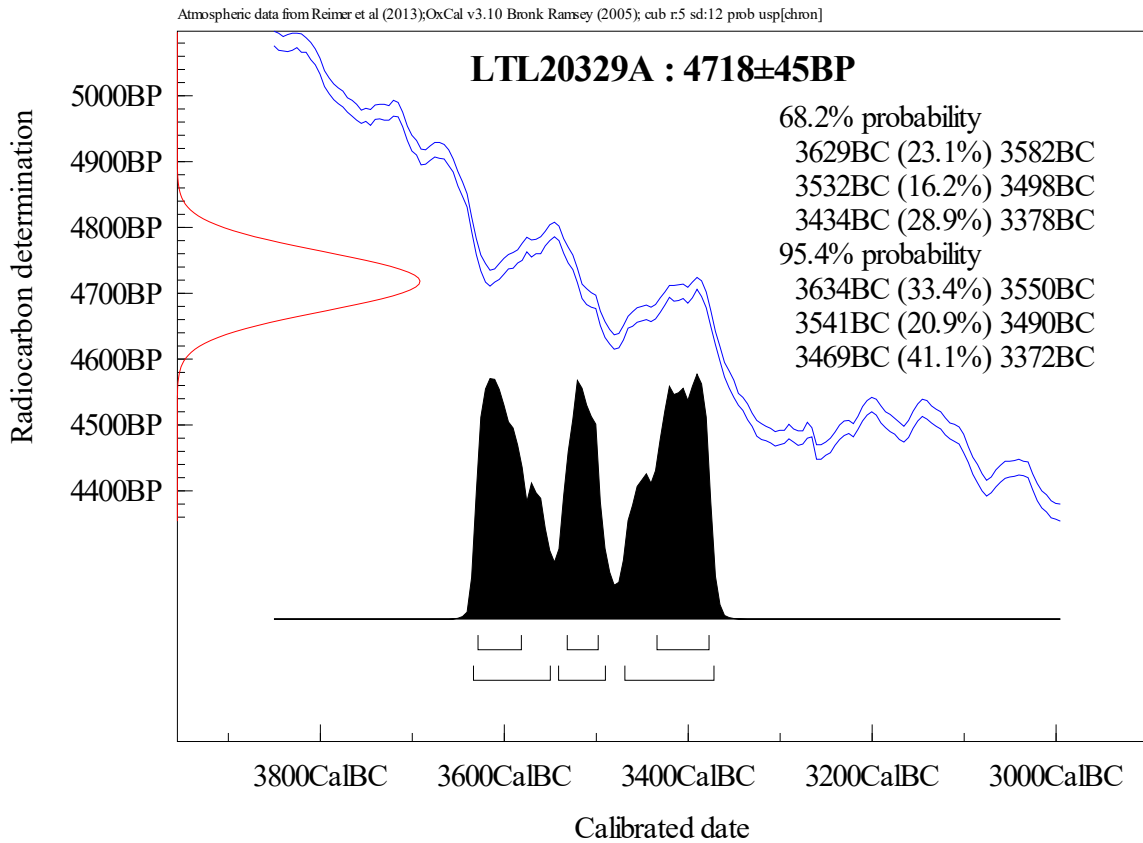
The "conventional radiocarbon age" was calculated with a  $\delta^{13}\text{C}$  correction based on the  $^{13}\text{C}/^{12}\text{C}$  ratio measured directly with the accelerator. For the estimation of the measurement uncertainty (standard deviation) both the radioisotope counting statistics and the scattering of the data have been taken into account. The larger of the two is given as final error in Table 2.

<b>Sample</b>	<b>Radiocarbon Age (BP)</b>	<b><math>\delta^{13}\text{C}</math> (‰)<sup>(**)</sup></b>	<b>Note</b>
<b>LTL20328A</b>	<b>2170 ± 45</b>	<b>-27.0 ± 0.4</b>	
<b>LTL20329</b>	<b>4718 ± 45</b>	<b>-30.2 ± 0.6</b>	
<b>LTL20330A</b>	<b>2685 ± 45</b>	<b>-26.5 ± 0.6</b>	
<b>LTL20331A</b>	<b>2185 ± 45</b>	<b>-24.8 ± 0.6</b>	
<b>LTL20332A</b>	<b>2017 ± 45</b>	<b>-24.5 ± 0.2</b>	
<b>LTL20333A</b>	<b>2484 ± 45</b>	<b>-28.2 ± 0.4</b>	

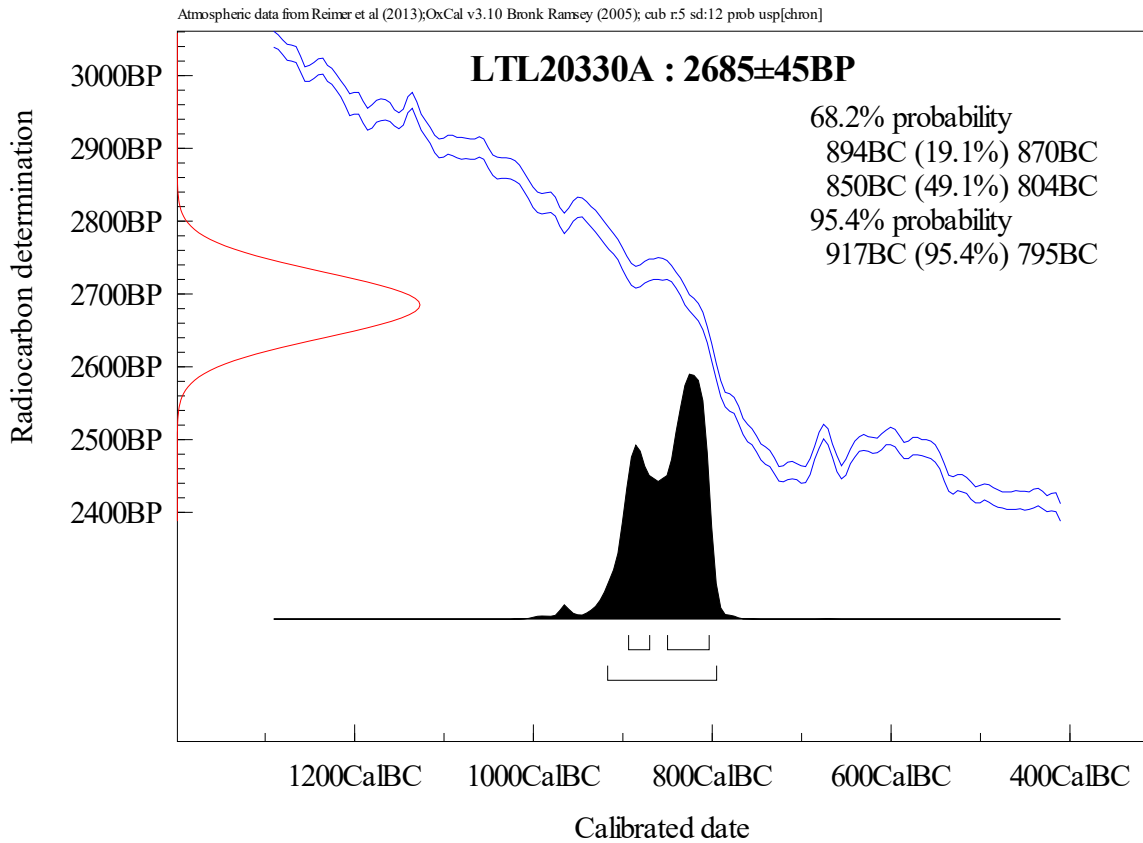
(\*\*) The listed values of the carbon stable isotopes fractionation term ( $\delta^{13}\text{C}$ ) are measured by AMS. These values can differ from the natural fractionation and from those measured by IRMS.

The conventional radiocarbon ages of the samples were converted into calendar years by using the software OxCal Ver. 3.5 based on the last atmospheric dataset [Reimer PJ, et al. 2013 *Radiocarbon* 55 No. 4-1869-1887]. The results of the calibration are reported in the following figures.

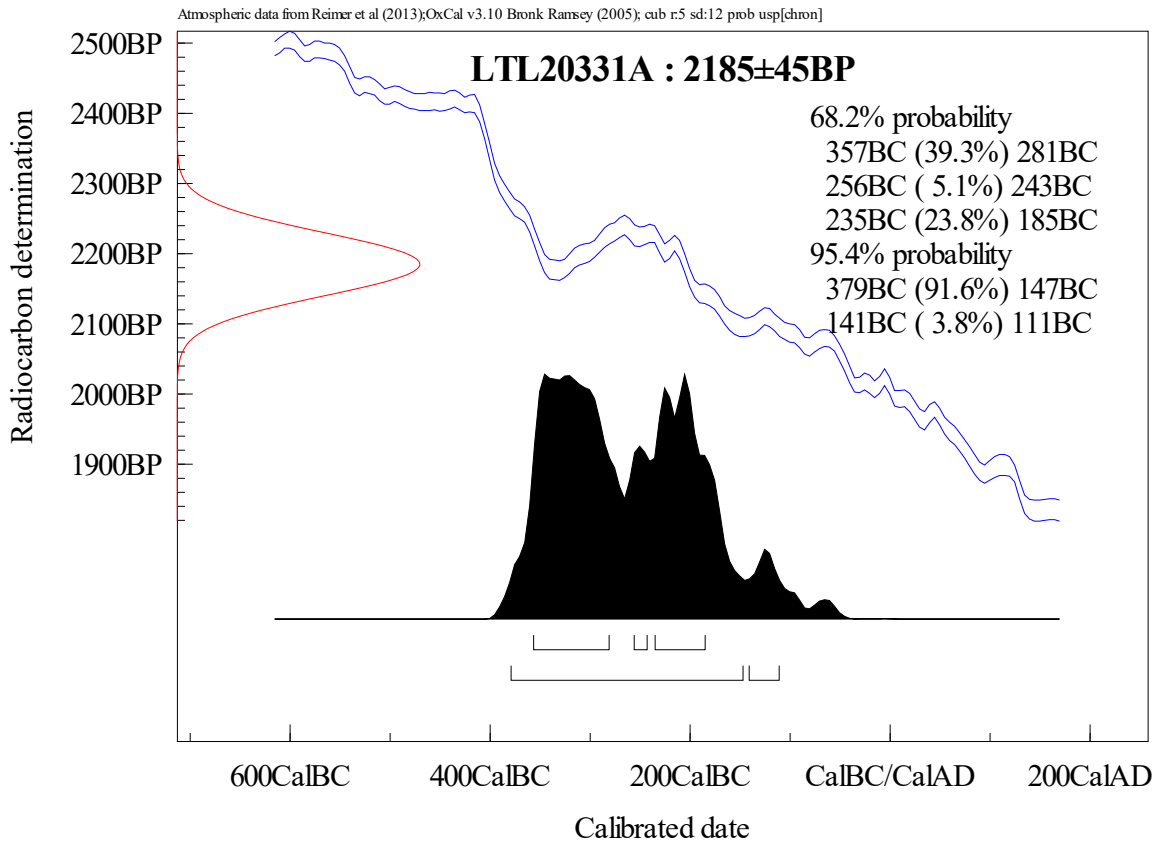




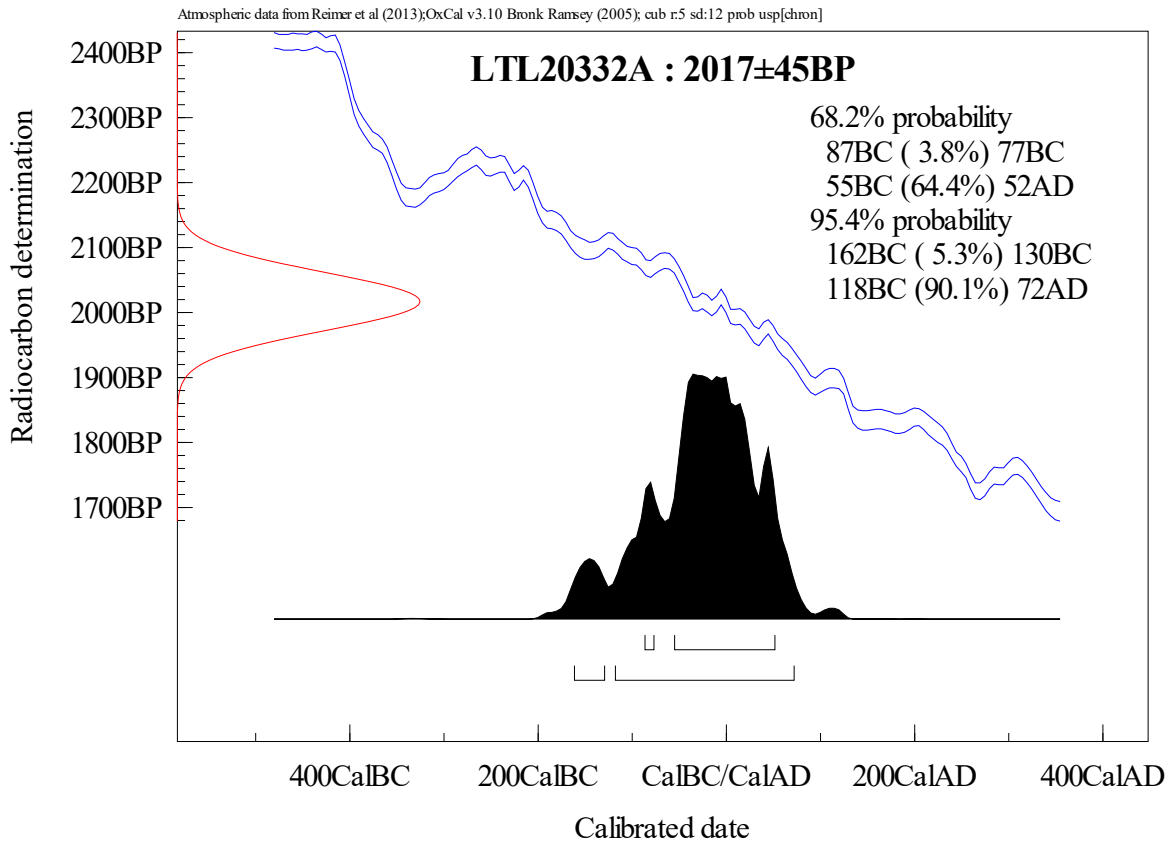
**Figure 2. Calibration of the radiocarbon age of the sample LTL20329A.**



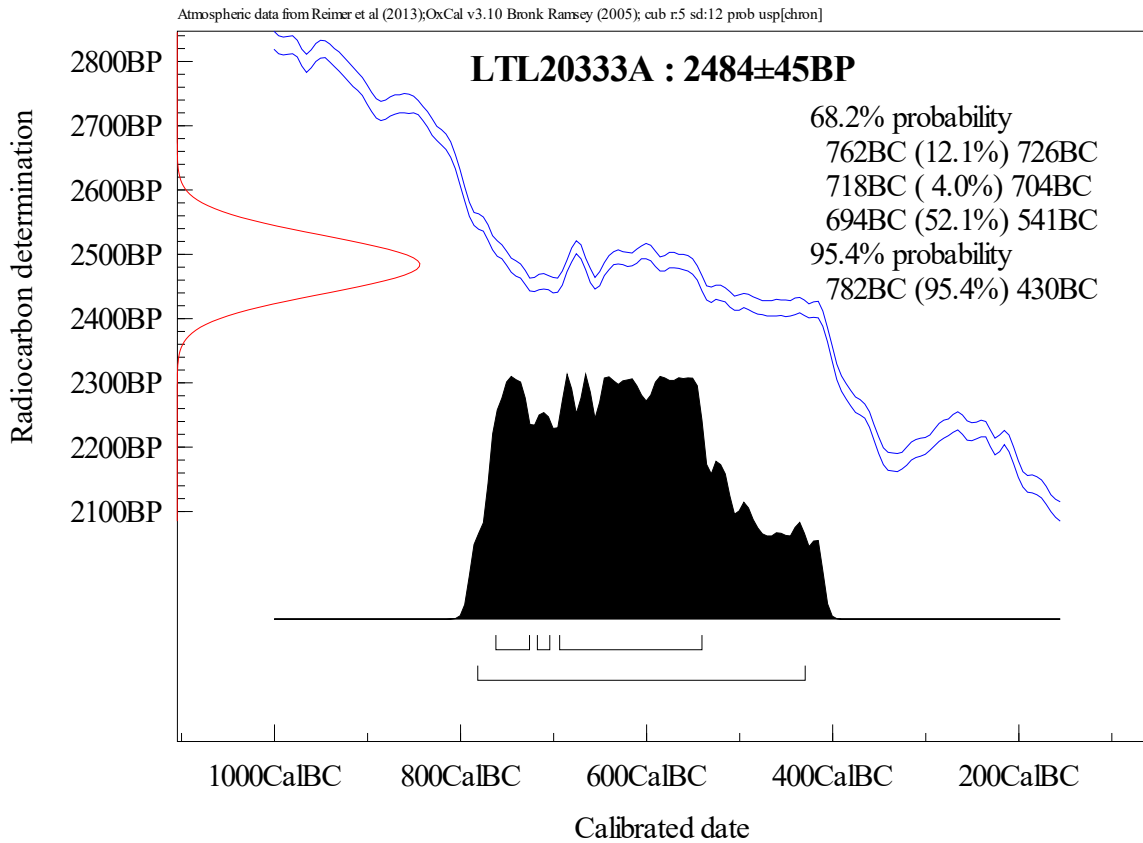
**Figure 3. Calibration of the radiocarbon age of the sample LTL20330A.**



**Figure 4. Calibration of the radiocarbon age of the sample LTL20331A.**



**Figure 5. Calibration of the radiocarbon age of the sample LTL20332A.**



**Figure 6. Calibration of the radiocarbon age of the sample LTL20333A.**

Best Regards,

Prof. Dr. Lucio Calcagnile

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